

## The Construction of Chiral Fused Azabicycles Using a Pd-Catalyzed Allylic Substitution Cascade and Asymmetric Desymmetrization Strategy

Qianjin An, Delong Liu, \*, Iiefeng Shen, Yangang Liu, and Wanbin Zhang, Iiefeng Shen, Yangang Liu,

<sup>†</sup>School of Pharmacy and <sup>‡</sup>School of Chemistry and Chemical Engineering, Shanghai Jiao Tong University, 800 Dongchuan Road, Shanghai 200240, P. R. China

Supporting Information

ABSTRACT: A highly enantioselective Pd-catalyzed asymmetric allylic substitution cascade of cyclic N-sulfonylimines with an accompanying asymmetric desymmetrization has been developed for the construction of fused tetrahydroindole derivatives bearing two chiral centers. Mechanistic studies confirmed that the cascade reaction proceeds by initial allylic alkylation and subsequent allylic amination. The first alkylation is a chirality-control step and represents an asymmetric desymmetrization of ciscyclic allyl diacetates. The reaction has been performed on a gram scale, and the desired products can take part in several transformations.

hiral fused azabicycles represent an important class of compounds because of their prevalence in bioactive natural products and chiral drugs. Among these, azabicycles possessing octahydroindole skeletons are some of the most important.<sup>2</sup> For example, such compounds can serve as key intermediates in the synthesis of angiotensin-converting enzyme inhibitors (ACEIs) such as trandolapril and perindopril.<sup>3</sup> Several strategies for the synthesis of this skeleton have been reported, but these methodologies can only be used for the synthesis of a small variety of azabicyclic products using a limited number of substrates. Therefore, an efficient pathway toward the synthesis of diverse chiral fused azabicyclic octahydroindoles from simple and abundant starting materials remains challenging.

Pd-catalyzed allylic substitution is a powerful synthetic tool for the formation of C-C and C-X bonds (X = N, O, S, etc.). Asymmetric cascade reactions that utilize allylic substitutions provide an efficient pathway for the construction of heterocycles.<sup>6,7</sup> Two strategies utilizing asymmetric cascade reactions based on Pd-catalyzed allylic substitutions have been adopted for the construction of heterocycles. One utilizes a double asymmetric allylic substitution, which allows only the synthesis of single heterocycles bearing a few chiral centers, with somewhat modest catalytic behavior in most cases.<sup>6</sup> A different methodology utilizes an asymmetric conjugate addition-allylic substitution cascade, which also gives single heterocyclic products but with multiple chiral centers and excellent stereoselectivities.7

Saccharin-derived cyclic N-sulfonylimines are an intriguing class of synthons and have been used in several asymmetric reactions such as hydrogenations, nucleophilic additions, and cycloadditions. 10 N-Sulfonylimines have also been used as binucleophiles for the construction of chiral piperidine rings in metal-free asymmetric cascade reactions (sequential C- and Nalkylations) (Scheme 1a). 11 On the basis of our previous work concerning metal-catalyzed asymmetric allylic substitutions 12 and the use of N-sulfonylimines as binucleophiles in organocatalyzed asymmetric cascade reactions, <sup>13</sup> we herein report the efficient construction of fused chiral octahydroindole derivatives bearing multiple chiral centers. The strategy employs Nsulfonylimines as binucleophiles and cis-cyclic allyl diacetates as allylic substrates 14 in a Pd-catalyzed asymmetric allylic substitution cascade with an accompanying asymmetric desymmetrization (Scheme 1b).

Initially, we carried out the Pd-catalyzed asymmetric allylic substitution cascade of cyclic N-sulfonylimine 1a and ciscyclohex-2-ene-1,4-diyl diacetate (2a) for screening of the reaction conditions. After examination of several chiral ligands, we found that our previously developed <sup>t</sup>Bu-RuPHOX<sup>15</sup> ligand is the best one for the reaction. <sup>16</sup> The effects of different Pd sources, solvents, and bases on the cascade reaction were then examined. 16 The optimal reaction conditions were found to be a catalyst system consisting of  $[Pd(\eta^3-C_3H_5)Cl]_2$  and  ${}^tBu$ RuPHOX in the presence of DBU in 1,4-dioxane under a N<sub>2</sub> atmosphere at 25 °C (Scheme 2).

Next, the applicability of the asymmetric allylic substitution cascade reaction of 1 bearing different R<sup>1</sup> and R<sup>2</sup> substituents

Received: November 26, 2016 Published: December 23, 2016



Organic Letters Letter

## Scheme 1. Use of N-Sulfonylimines in Asymmetric Catalysis

(b) This work (as binuclophiles in metal-mediated asymmetric catalysis)

$$R^2$$
 $R^1$ 
 $QAC$ 
 $Pd/L^*$ 
 $R^2$ 
 $R^1$ 
 $R^2$ 
 $R^2$ 
 $R^3$ 
 $R^4$ 
 $R^2$ 
 $R^3$ 
 $R^4$ 
 $R^4$ 

Scheme 2. Screening of the Reaction Conditions

was evaluated (Scheme 3). N-Sulfonylimines with an electronwithdrawing group located on the phenyl ring were first investigated. Similar to 1a, all of these substrates (1b-h) provided the desired products in high yields with excellent diastereo- and enantioselectivities. We were pleased to discover that the asymmetric catalytic results for the above reactions were unaffected by steric hindrance. The reaction of 1i containing two F atoms at the 3- and 4-positions of the phenyl ring also gave excellent catalytic results. N-Sulfonylimines 1j-n bearing electron-donating substituents were also examined. The desired products were obtained in yields in excess of 90% with excellent enantioselectivities. When the phenyl ring was replaced by a naphthalene (10), excellent catalytic results were also observed. Finally, the aromatic ring of the Nsulfonylimines was replaced by aliphatic substituents (1p-r). To our delight, these substrates all provided excellent enantioselectivities, albeit in moderate yields, with 3r bearing a Ph group being obtained with 99.8% ee. We also carried out the asymmetric cascade reaction by using a racemic transcycloalkenediol diacetate instead of cis-isomer 2a. 17 Full conversion but complicated products were obtained even after 72 h.16

Substrates 1s and 1t with different  $R^2$  substituents (Me and t-Bu, respectively) were next examined. High yields and excellent enantioselectivities were still obtained for the cascade reaction. In addition, a high yield but moderate enantioselectivity was obtained when a five-membered cyclic diacetate (2u) was used. However, the use of a seven-membered cyclic diacetate (2v) gave the corresponding product with high enantioselectivity but low yield.

To evaluate the practicality of this catalytic process, a gramscale reaction of **1a** with **2a** was carried out under the optimal reaction conditions. As shown in Scheme 4, the desired product

Scheme 3. Substrate Scope<sup>a</sup>

"Using the optimal reaction conditions shown in Scheme 2 with <sup>t</sup>Bu-RuPHOX as the ligand in 1,4-dioxane. All of the reactions afforded >20:1 dr. Isolated yields are shown. The ee values were determined by HPLC using chiral Daicel columns.

3a was obtained in 92% yield with 94% ee. 3a has the potential to participate in a variety of transformations. First, the SO<sub>2</sub> group of 3a and the double bond of the enamide were removed and reduced simultaneously using sodium naphthalide (4, pathway a). 10a The double bond of the cyclohexene ring can also be hydrogenated using RuCl<sub>2</sub>(PPh<sub>3</sub>)<sub>3</sub> as a catalyst under 40 bar H<sub>2</sub> at room temperature (5, pathway b). When 3a is treated with H<sub>2</sub> and a Pd/C catalyst (pathway c), the above two double bonds can be reduced simultaneously to afford octahydroindole 6 in 99% yield with 1:1 dr. Alternatively, the double bond of the enamide can be reduced in the presence of Et<sub>3</sub>SiH/BF<sub>3</sub>·Et<sub>2</sub>O to give 7 (pathway d),13c which can then be subjected to hydrogenation (pathway c) for 2 h to give octahydroindole 8 in 80% yield with 95% ee after a simple recrystallization. Finally, the double bond of the enamide can be oxidized regioselectively by treating 3a with m-CPBA, affording the oxidized product 9 in high yield and ee; 9 is a key intermediate for the synthesis of chiral alicyclic amino alcohols that possess pharmacological properties.<sup>18</sup>

In order to determine the reaction pathway, the asymmetric allylic substitution cascade of 1a and 2a was carried out using the above optimal reaction conditions at  $-30\,^{\circ}\text{C}$  for 24 h, with the exception that THF was used in place of 1,4-dioxane

Organic Letters Letter

# Scheme 4. Large-Scale Synthesis of 3a and Its Transformations<sup>a</sup>

<sup>a</sup>Conditions: (a) Sodium naphthalide, DME, −78 °C to rt, 2 h; Ac<sub>2</sub>O, DMAP, DCM, 4 h. (b) RuCl<sub>2</sub>(PPh<sub>3</sub>)<sub>3</sub>, K<sub>2</sub>CO<sub>3</sub>, CH<sub>3</sub>OH, 40 bar H<sub>2</sub>, 12 h. (c) Pd/C, CH<sub>3</sub>OH, H<sub>2</sub> balloon, 12 h. (d) Et<sub>3</sub>SiH/BF<sub>3</sub>·Et<sub>2</sub>O, DCM, −20 °C, 24 h. (e) *m*-CPBA, DCM, −20 °C, 8 h.

(Scheme 5). A mixture of diastereomers 10 was obtained as the major product with only a trace amount of 3a and recovery of

#### Scheme 5. Reaction Pathway

the remaining starting material. Without further isolation, the mixture of 10 was subjected to further transformations using racemic BINAP as a ligand, providing the desired product 3a in 98% yield with 96% ee.

The above results suggested that the fused azacycle 3a is constructed via an allylic alkylation followed by an allylic amination. The reaction pathway is envisaged to proceed as follows (Scheme 6). First, combination of cis-2a with  $L_2Pd^0$  provides allyl-Pd complex A. As expected, the process represents an asymmetric desymmetrization of cis-2a because  $L_2Pd^0$  prefers to attack the R-chiral carbon of 2a. Complex A then reacts with nucleophile 1a to give alkylated intermediate B. Finally, B takes part in the next allylic amination, giving the terminal fused heterocycle 3a in high yield and enantioselectivity via allyl-Pd complex C. The original catalyst system  $L_2Pd^0$  is subsequently regenerated. It is clear that the asymmetric desymmetrization is the chirality-control step and that the chirality of 3a is determined by B.

In summary, we have developed an efficient pathway for the construction of chiral fused azabicycles using a Pd-catalyzed allylic substitution cascade and asymmetric desymmetrization strategy. Under the optimal reaction conditions, a series of cyclic *N*-sulfonylimines can be used to give the desired products

#### Scheme 6. Proposed Reaction Mechanism

in high yields with up to 99.8% ee. The reaction was performed on a gram scale, and the corresponding products allowed for several transformations. Mechanistic studies confirmed that the cascade reaction undergoes an allylic alkylation followed by an allylic amination. The first alkylation is the chirality-control step and represents an asymmetric desymmetrization of *cis-2*. The alkylated intermediates determine the configuration of the terminal cascade products 3 via a subsequent amination step.

### ASSOCIATED CONTENT

#### S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b03529.

Crystallographic data for 3a (CIF)

Crystallographic data for 3q (CIF)

Crystallographic data for 6-2 (CIF)

Crystallographic data for 8 (CIF)

Crystallographic data for 9 (CIF)

Experimental procedures and characterization data for all reactions and products, including <sup>1</sup>H and <sup>13</sup>C NMR spectra, HPLC data, and crystal data (PDF)

## **■** AUTHOR INFORMATION

#### **Corresponding Authors**

\*E-mail: dlliu@sjtu.edu.cn. \*E-mail: wanbin@sjtu.edu.cn.

ORCID ®

Wanbin Zhang: 0000-0002-4788-4195

#### Notes

The authors declare no competing financial interest.

## ACKNOWLEDGMENTS

This work was partially supported by the National Natural Science Foundation of China (21232004, 21372152, 21402117, 21472123, and 21672142), the Program of Shanghai Subject Chief Scientists (14XD1402300), and the Instrumental Analysis Center of SJTU for characterization.

Organic Letters Letter

### **■** REFERENCES

- (1) For selected reviews of chiral fused azabicycles, see: (a) Migliori, G. B.; Dheda, K.; Centis, R.; Mwaba, P.; Bates, M.; O'Grady, J.; Hoelscher, M.; Zumla, A. *Trop. Med. Int. Health* **2010**, *15*, 1052. (b) Drawz, S. M.; Bonomo, R. A. *Clin. Microbiol. Rev.* **2010**, *23*, 160.
- (2) For selected reviews, see: (a) Ersmark, K.; Del Valle, J. R.; Hanessian, S. Angew. Chem., Int. Ed. 2008, 47, 1202. (b) Sayago, F. J.; Laborda, P.; Calaza, M. I.; Jiménez, A. I.; Cativiela, C. Eur. J. Org. Chem. 2011, 2011, 2011.
- (3) For selected papers, see: (a) Singh, G. P.; Godbole, H. M.; Nehate, S. P.; Mahajan, P. R. Synth. Commun. 2005, 35, 243. (b) Yeki, M.; Koda, M.; Matono, T.; Sugihara, T.; Maeda, K.; Murawaki, Y. Mol. Med. Rep. 2009, 2, 857. (c) Bhattacharya, A.; Chattopadhyay, B.; Chakraborty, S.; Roy, B. N.; Singh, G. P.; Godbole, H. M.; Rananaware, U. B.; Mukherjee, A. K. J. Pharm. Biomed. Anal. 2012, 70, 280. (d) Gomez, C.; Berteina-Raboin, S.; De Nanteuil, G.; Guillaumet, G. Bioorg. Med. Chem. 2013, 21, 7216. (e) Qabaja, G.; Benavides, A. R.; Liu, S.; Petersen, K. S. J. Org. Chem. 2015, 80, 133. (4) For recent papers, see: (a) Hesp, K. D.; Tobisch, S.; Stradiotto, M. J. Am. Chem. Soc. 2010, 132, 413. (b) Clarisse, D.; Fenet, B.; Fache, F. Org. Biomol. Chem. 2012, 10, 6587. (c) Aron, Z. D.; Ito, T.; May, T. L.; Overman, L. E.; Wang, J. J. Org. Chem. 2013, 78, 9929. (d) Musacchio, A. J.; Nguyen, L. Q.; Beard, G. H.; Knowles, R. R. J. Am. Chem. Soc. 2014, 136, 12217. (e) Han, Y.; Zheng, B.; Peng, Y. Adv. Synth. Catal. 2015, 357, 1136. (f) Wappes, E. A.; Fosu, S. C.; Chopko, T. C.; Nagib, D. A. Angew. Chem., Int. Ed. 2016, 55, 9974. (g) Hazelden, I. R.; Ma, X.; Langer, T.; Bower, J. F. Angew. Chem., Int. Ed. 2016, 55, 11198.
- (5) For selected reviews of Pd-catalyzed allylic substitutions, see: (a) Trost, B. M.; Van Vranken, D. L. Chem. Rev. 1996, 96, 395. (b) Helmchen, G.; Pfaltz, A. Acc. Chem. Res. 2000, 33, 336. (c) Trost, B. M.; Crawley, M. L. Chem. Rev. 2003, 103, 2921. (d) Lu, Z.; Ma, S. Angew. Chem., Int. Ed. 2008, 47, 258. (e) Trost, B. M.; Zhang, T.; Sieber, J. D. Chem. Sci. 2010, 1, 427. (f) Trost, B. M. Org. Process Res. Dev. 2012, 16, 185. (g) Lumbroso, A.; Cooke, M. L.; Breit, B. Angew. Chem., Int. Ed. 2013, 52, 1890. (h) Butt, N.; Zhang, W. Chem. Soc. Rev. 2015, 44, 7929. (i) Butt, N.; Yang, G.; Zhang, W. Chem. Rec. 2016, 16, 2683.
- (6) For selected papers, see: (a) Uozumi, Y.; Tanahashi, A.; Hayashi, T. J. Org. Chem. 1993, 58, 6826. (b) Gaucher, A.; Dorizon, P.; Ollivier, J.; Salaün, J. Tetrahedron Lett. 1995, 36, 2979. (c) Yoshida, M.; Maeyama, Y.; Shishido, K. Tetrahedron Lett. 2010, 51, 6008. (d) Ye, K.-Y.; He, H.; Liu, W.-B.; Helmchen, G.; Dai, L.-X.; You, S.-L. J. Am. Chem. Soc. 2011, 133, 19006. (e) Bartlett, M. J.; Turner, C. A.; Harvey, J. E. Org. Lett. 2013, 15, 2430. (f) Yoshida, M.; Kinoshita, K.; Namba, K. Org. Biomol. Chem. 2014, 12, 2394.
- (7) For selected papers, see: (a) Shintani, R.; Park, S.; Shirozu, F.; Murakami, M.; Hayashi, T. J. Am. Chem. Soc. 2008, 130, 16174. (b) Trost, B. M.; Bringley, D. A.; Silverman, S. M. J. Am. Chem. Soc. 2011, 133, 7664. (c) Lowe, M. A.; Ostovar, M.; Ferrini, S.; Chen, C. C.; Lawrence, P. G.; Fontana, F.; Calabrese, A. A.; Aggarwal, V. K. Angew. Chem., Int. Ed. 2011, 50, 6370. (d) Williams, F. J.; Jarvo, E. R. Angew. Chem., Int. Ed. 2011, 50, 4459. (e) Wang, L.; Menche, D. Angew. Chem., Int. Ed. 2012, 51, 9425. (f) Trost, B. M.; Bringley, D. A. Angew. Chem., Int. Ed. 2013, 52, 4466. (g) Khan, A.; Yang, L.; Xu, J.; Jin, L. Y.; Zhang, Y. J. Angew. Chem., Int. Ed. 2014, 53, 11257. (h) Ohmatsu, K.; Imagawa, N.; Ooi, T. Nat. Chem. 2014, 6, 47. (i) Xu, C.-F.; Zheng, B.-H.; Suo, J.-J.; Ding, C.-H.; Hou, X.-L. Angew. Chem., Int. Ed. 2015, 54, 1604.
- (8) (a) Chen, Q.-A.; Ye, Z.-S.; Duan, Y.; Zhou, Y.-G. Chem. Soc. Rev. 2013, 42, 497 and references cited therein. (b) Wang, L.; Zhou, Q.; Qu, C. H.; Wang, Q. W.; Cun, L. F.; Zhu, J.; Deng, J. G. Tetrahedron 2013, 69, 6500. (c) Sugie, H.; Hashimoto, Y.; Haraguchi, N.; Itsuno, S. J. Organomet. Chem. 2014, 751, 711.
- (9) (a) Luo, Y.; Carnell, A. J.; Lam, H. W. Angew. Chem., Int. Ed. 2012, 51, 6762. (b) Nishimura, T.; Noishiki, A.; Tsui, G. C.; Hayashi, T. J. Am. Chem. Soc. 2012, 134, 5056. (c) Wang, H.; Jiang, T.; Xu, M.-H. J. Am. Chem. Soc. 2013, 135, 971. (d) Yang, G.; Zhang, W. Angew. Chem., Int. Ed. 2013, 52, 7540. (e) Jiang, C.; Lu, Y.; Hayashi, T. Angew.

Chem., Int. Ed. 2014, 53, 9936. (f) Chen, Y.-J.; Chen, Y.-H.; Feng, C.-G.; Lin, G.-Q. Org. Lett. 2014, 16, 3400.

- (10) (a) Rommel, M.; Fukuzumi, T.; Bode, J. W. J. Am. Chem. Soc. 2008, 130, 17266. (b) Chiang, P.-C.; Rommel, M.; Bode, J. W. J. Am. Chem. Soc. 2009, 131, 8714. (c) Zhu, B.-H.; Zheng, J.-C.; Yu, C.-B.; Sun, X.-L.; Zhou, Y.-G.; Shen, Q.; Tang, Y. Org. Lett. 2010, 12, 504. (d) Chen, X.-Y.; Lin, R.-C.; Ye, S. Chem. Commun. 2012, 48, 1317. (e) Feng, X.; Zhou, Z.; Ma, C.; Yin, X.; Li, R.; Dong, L.; Chen, Y.-C. Angew. Chem., Int. Ed. 2013, 52, 14173. (f) Nishimura, T.; Ebe, Y.; Hayashi, T. J. Am. Chem. Soc. 2013, 135, 2092. (g) Yin, X.; Zheng, Y.; Feng, X.; Jiang, K.; Wei, X.-Z.; Gao, N.; Chen, Y.-C. Angew. Chem., Int. Ed. 2014, 53, 6245. (h) Fei, J.; Qian, Q.; Sun, X.; Gu, X.; Zou, C.; Ye, J. Org. Lett. 2015, 17, 5296.
- (11) (a) Xiong, X.-F.; Zhang, H.; Peng, J.; Chen, Y.-C. Chem. Eur. J. 2011, 17, 2358. (b) Kravina, A. G.; Mahatthananchai, J.; Bode, J. W. Angew. Chem., Int. Ed. 2012, 51, 9433. (c) Chen, X.-Y.; Gao, Z.-H.; Song, C.-Y.; Zhang, C.-L.; Wang, Z.-X.; Ye, S. Angew. Chem., Int. Ed. 2014, 53, 11611. (d) Gu, J.; Ma, C.; Li, Q.-Z.; Du, W.; Chen, Y.-C. Org. Lett. 2014, 16, 3986.
- (12) For selected recent papers, see: (a) Zhao, X.; Liu, D.; Guo, H.; Liu, Y.; Zhang, W. J. Am. Chem. Soc. 2011, 133, 19354. (b) Quan, M.; Butt, N.; Shen, J.; Shen, K.; Liu, D.; Zhang, W. Org. Biomol. Chem. 2013, 11, 7412. (c) Huo, X.; Quan, M.; Yang, G.; Zhao, X.; Liu, D.; Liu, Y.; Zhang, W. Org. Lett. 2014, 16, 1570. (d) Huo, X.; Yang, G.; Liu, D.; Liu, Y.; Gridnev, I. D.; Zhang, W. Angew. Chem., Int. Ed. 2014, 53, 6776. (e) Wei, X.; Liu, D.; An, Q.; Zhang, W. Org. Lett. 2015, 17, 5768. (f) Huo, X.; He, R.; Zhang, X.; Zhang, W. J. Am. Chem. Soc. 2016, 138, 11093.
- (13) (a) An, Q.; Shen, J.; Butt, N.; Liu, D.; Liu, Y.; Zhang, W. Org. Lett. 2014, 16, 4496. (b) An, Q.; Li, J.; Shen, J.; Butt, N.; Liu, D.; Liu, Y.; Zhang, W. Chem. Commun. 2015, 51, 885. (c) An, Q.; Shen, J.; Butt, N.; Liu, D.; Liu, Y.; Zhang, W. Adv. Synth. Catal. 2015, 357, 3627.
- (14) This type of substrate has been used in Pd-catalyzed asymmetric allylic substitutions. See: (a) Trost, B. M.; Li, L.; Guile, S. D. J. Am. Chem. Soc. 1992, 114, 8745. (b) Yoshizaki, H.; Satoh, H.; Sato, Y.; Nukui, S.; Shibasaki, M.; Mori, M. J. Org. Chem. 1995, 60, 2016. (c) Trost, B. M.; Surivet, J.-P. Angew. Chem., Int. Ed. 2000, 39, 3122. (d) Chapsal, B. D.; Ojima, I. Org. Lett. 2006, 8, 1395.
- (15) For reviews, see: (a) Zhang, W.; Liu, D. In Chiral Ferrocenes in Asymmetric Catalysis: Synthesis and Applications; Dai, L.-X., Hou, X.-L., Eds.; Wiley-VCH: Weinheim, Germany, 2010; Chapter 14, pp 175–214. (b) Butt, N.; Liu, D.; Zhang, W. Synlett 2014, 25, 615. For papers, see: (c) Zhang, W.; Adachi, Y.; Hirao, T.; Ikeda, I. Tetrahedron: Asymmetry 1996, 7, 451. (d) Liu, D.; Xie, F.; Zhang, W. Tetrahedron Lett. 2007, 48, 585. (e) Liu, D.; Xie, F.; Zhao, X.; Zhang, W. Tetrahedron 2008, 64, 3561.
- (16) See details in the Supporting Information.
- (17) Okauchi, T.; Fujita, K.; Ohtaguro, T.; Ohshima, S.; Minami, T. *Tetrahedron: Asymmetry* **2000**, *11*, 1397.
- (18) Nemoto, T.; Fukuyama, T.; Yamamoto, E.; Tamura, S.; Fukuda, T.; Matsumoto, T.; Akimoto, Y.; Hamada, Y. Org. Lett. 2007, 9, 927.